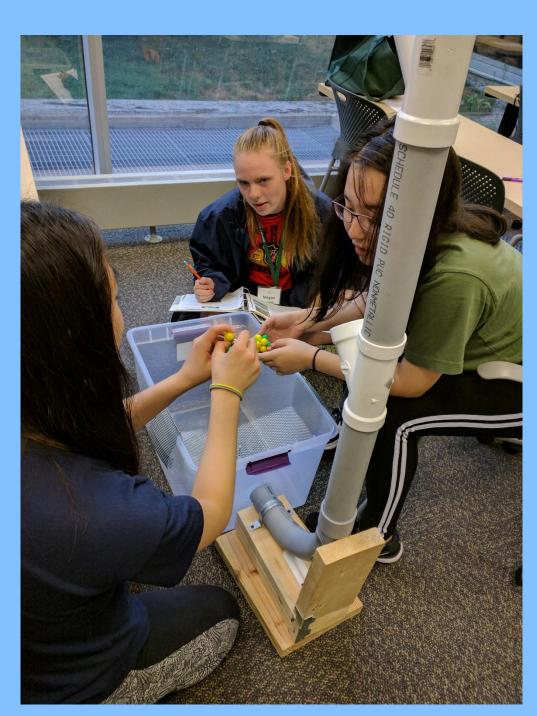


## Day 1 94 Hours Remaining





Modeling **Nuclear** Accelerators with Marbles

**Building a Lego Chart of** the Nuclide











**Students on** the Campus **Tour in Front** of the FRIB Facility

















# PANdamonum Megan Campbell, Alex Delp, Vidor Lujan

Day 2

### 70 Hours Remaining

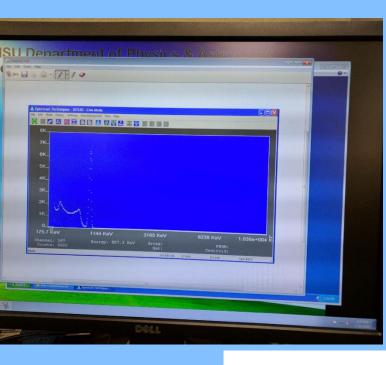
**Radiation Detectors Presentation and Q & A** 

#### **Gamma Spectroscopy Experiment**





The purpose of the lab is to discover the identity of an unknown isotope. We used a gamma ray energy detector and a multichannel analyzer and a computer program. Using commonly used gamma spectrum energy values for  $^{137}_{55}Cs$  and  $^{137}_{27}Co$ , we calibrated the program.



Then measured the energy the gamma rays of an unknown radioactive element. Using an online table and our peak energy value while taking into account the half life and the intensity/number of major peaks we determined our unknown was Mn-54





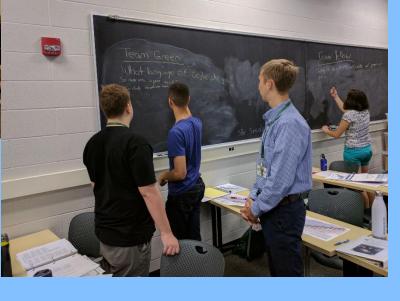
# Day 3 **46 Hours Remaining**





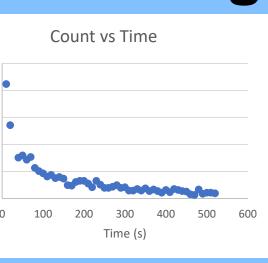
Neutron Capture Activity





#### **Measuring Nuclear Half-Life Labs**

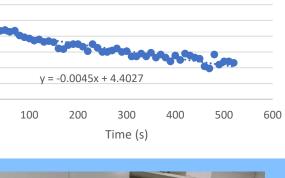
Campine Date

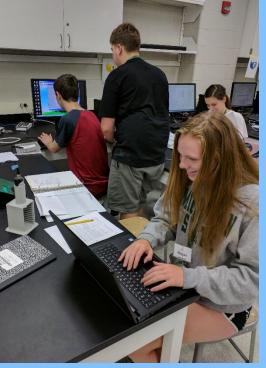


Q & A



In(count) vs time





The purpose of this lab was to find the half-life of a silver disk after it was put in proximity to a neutron source. We setup a Geiger-Muller counter and measured the  $\beta^-$  decay of the Ag disk.

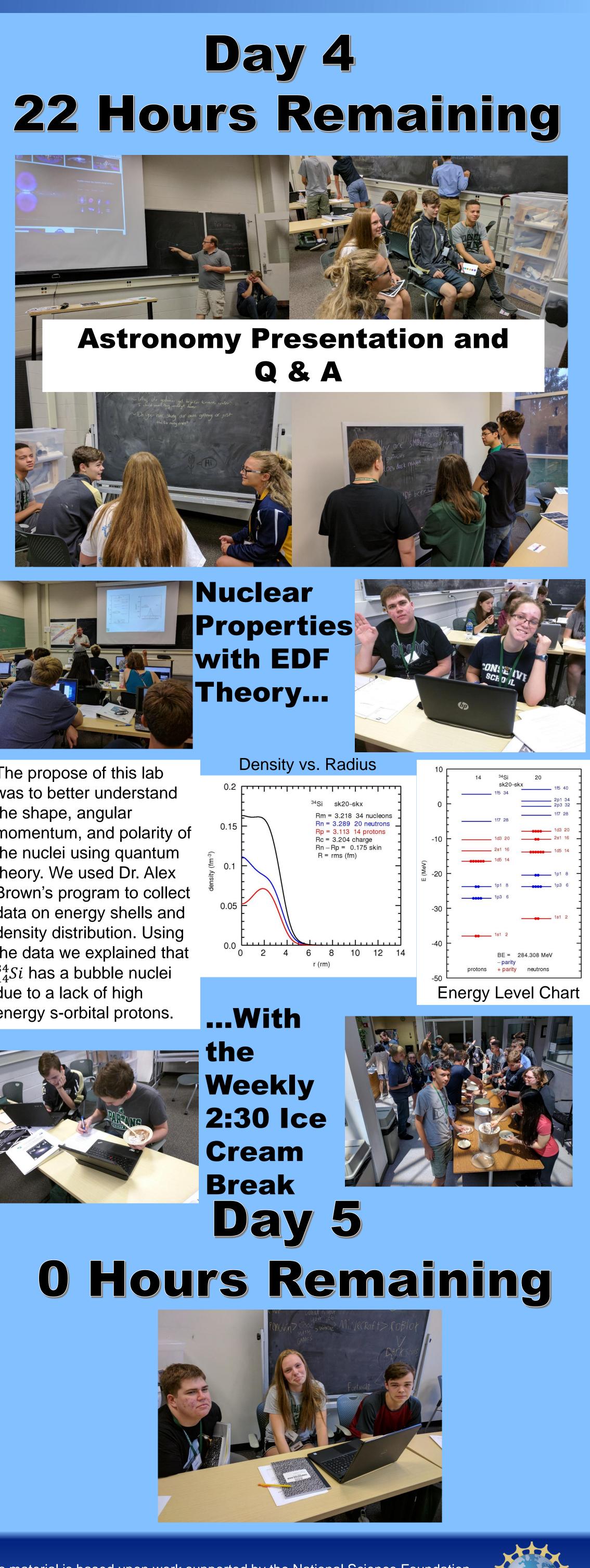
We created our count vs. time graph and because silver has a short half life, it was exponential. We linearized the graph to create our In(count vs. time graph.

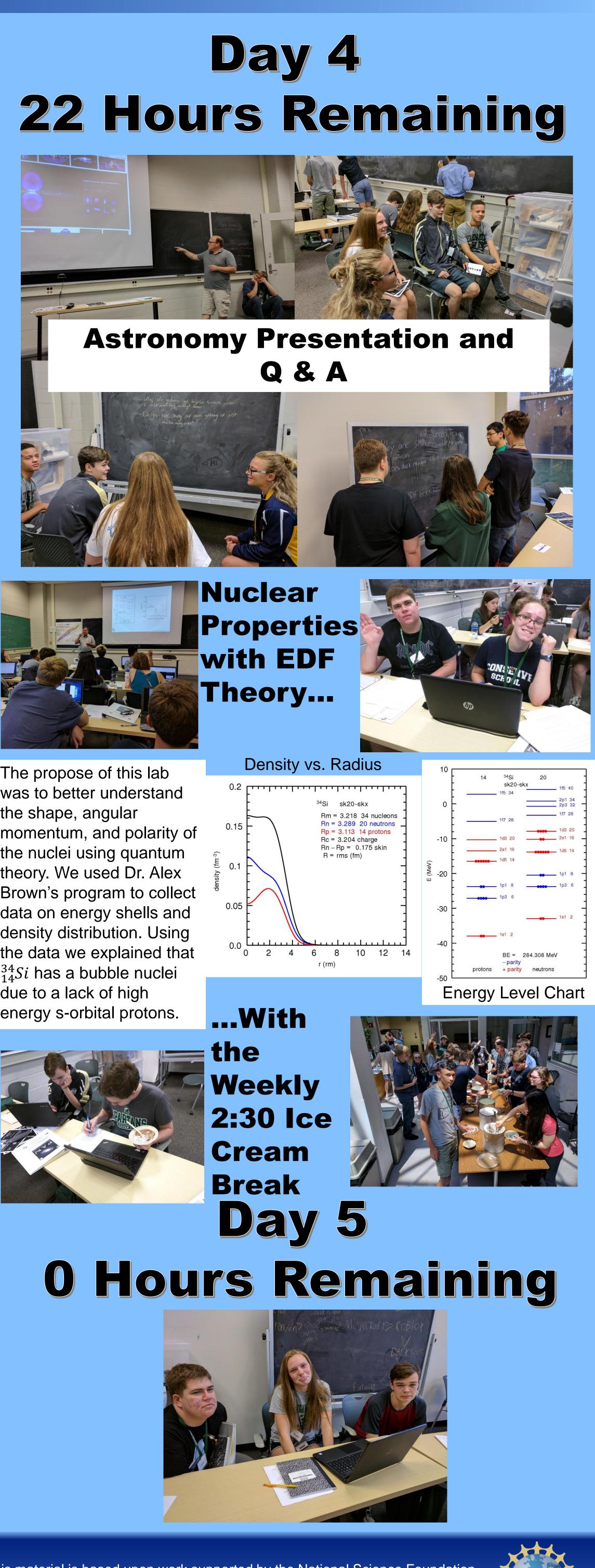
 $\ln(\text{count}) = -\ln(2)/t_{1/2}(\text{time}) + \text{initial count}$  $slope=-ln(2)/t_{1/2}$ 

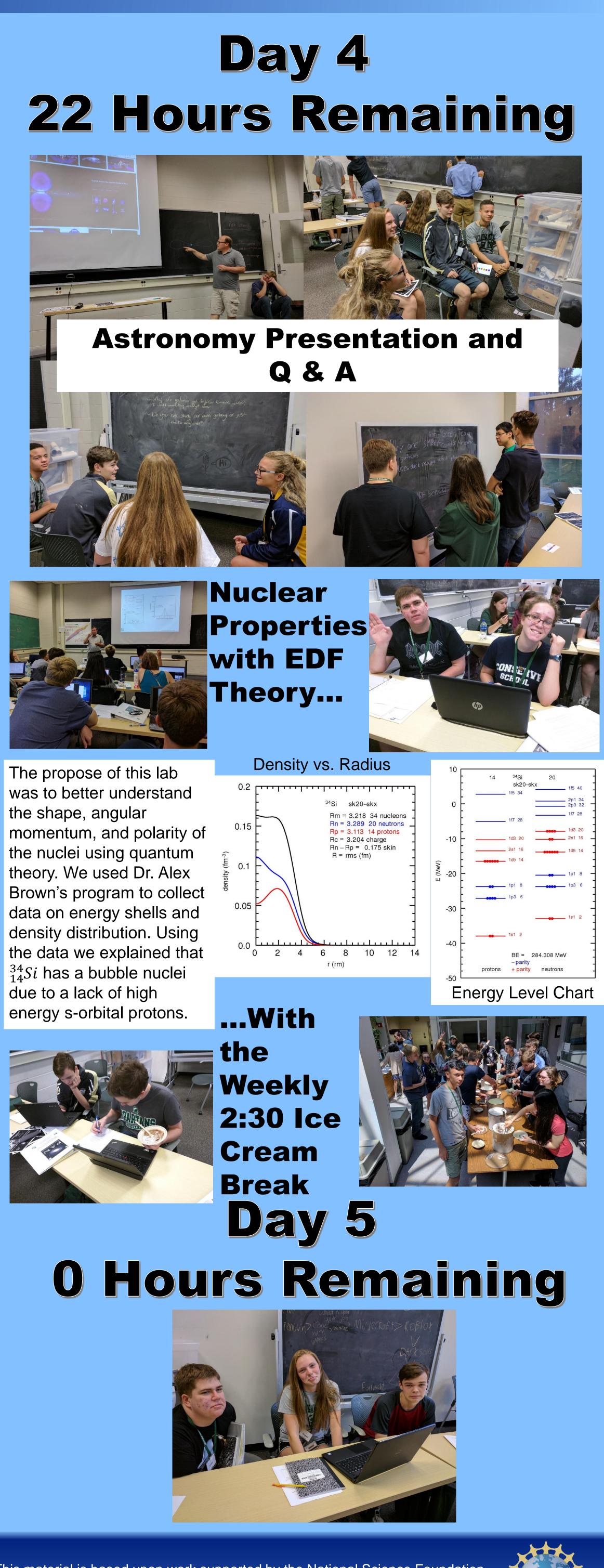
-0.0045=-ln(2)  $/t_{1/2}$ , so  $t_{1/2}$ =154 seconds The actual half-life is 142 seconds











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