Stars produce energy through fusion - the combining of light nuclei to make heavier ones. The "ashes" left over after fusion "burning" are new elements! While the Big Bang produced a lot of hydrogen, helium and a bit of lithium, all the heavier elements were made by nuclear reactions. **Stars are nucleus factories.**

Our Sun is currently fusing hydrogen nuclei (protons) to make helium. You can recreate this "proton-proton chain" process using small objects (pennies & nickels, for example) to represent protons and neutrons.

Follow these rules to model how a star fuses nuclei to make energy!

- 1. Start with 4 loose proton pieces to represent the hydrogen fuel in a star. You will also need two six-sided dice and two neutron pieces.
- 2. Perform the reactions below, following arrows from one to the next. If you make anything other than loose protons (H-2, He-3, He-4), keep it separate!
- 3. Once you make Helium-4, you win!



Of the four possible reactions above (numbered 1-4),

- Which reaction happened often?
- Which reactions were rare (difficult)?
- Why were those reactions rare (there are two different reasons)?

Because those reactions are difficult, the sun fuses hydrogen slowly and hasn't used it all up - that's a good thing for us.

Stellar Fusion: the p-p chain

How to Play

Big Bang Nucleosynthesis	Big Bang eosynthesisAccording to the Big Bang theory, about 14 billion years ago the universe was just a hot and dense soup of energy and particles (a plasma). As it expanded and cooled down, neutrons and protons were formed. After about 2 minutes the universe was cool enough so that protons and 	
How to Play		
	START: nearly all matter in the universe is protons $P = {}^{1}H = Hydrogen-1$	
	+ PN + N) + PN
	$PN = {}^{2}H = Hydrogen-2$	
	2 3 4 5-10	11 12
	$+ \begin{array}{c} P \\ N \\ - \end{array} + \begin{array}{c} P \\ - \end{array} + \begin{array}{c} N \\ - \end{array} + \begin{array}{c} P \\ - \end{array} + \begin{array}{c} N \\ - \end{array} + \begin{array}{c} P \\ - P \\ - P \\ - P \\ - \end{array} + P \\ - P \\ $	$\begin{array}{c} P \\ + P \\ - N \\ - P \\ \end{array}$
	PN = ³ H = Hydrogen-3 5-10 11	$\frac{P}{P} = {}^{3}He = Helium - 3$
		+ PN + N - P
	PIN = ⁴He = Helium-4	
	Count what's left: of the 10 protons (hydrogen are still hydrogen? How many helium-4 did y	n) you started with, how many you successfully make? Is your

Count what's left: of the 10 protons (hydrogen) you started with, how many are still hydrogen? How many helium-4 did you successfully make? Is your "universe" still mostly hydrogen, but with a little bit of helium? Try it again if you like to see how the results change or stay the same!